Calculations of the Fast Reactor, Tal Into Account The Value of Neutrons	dine -		SOV	/39-6	1-3-4/	/32
Table 3. Volume		( <b>Q</b> .)		(6)		(C)
Portions of the Basic Fissionable Isotope				11	111	(6)
$\epsilon_{\rm f}$ in Three Types of Reactors at $\kappa_{\rm eff} = 1$ .	·	ed :	 0,0512	0,0464	6,6300	
	1	ر يد)	0,0460	O 0400	0,0266	(10~- 14)
	ı	( <b>f</b> ,	 0,0540	0,0466	0,0300	<b>~</b> 1

Key to Table 3. (a) Method of computation; (b) reactor type; (c) difference from results of spatial computation, %; (d) nine group spacial; (e) single group without taking into account neutron values; (f) single group taking into account neutron values.

Card 9/9

NOVOZHILOV, A. I., Candidate Tech Sci (diss) -- "Investigation of the movement of a vibration machine for packing soil". Leningrad, 1959. 14 pp (Min Higher Educ USSR, Leningrad Order of Labor Red Banner Construction Engineering Inst Chair of "Theoretical Mechanics"), 150 copies (KL, No 24, 1959, 139)

# NOVOZHILOV, A.I., kand.tekhn.nauk

Impact speed of a vibration percussion machine. Izv.vys. ucheb.zav.; mashinostr. no.8:56-58 '62. (MIRA 15:12)

1. Magnitogorskiy gorno-metallurgicheskiy institut. (Machinery)

NOVOZHILOV, A.I., inzh.

Changing the direction of vibrations originated by eccentric vibrators. Stroi. i dor. mashinostr. no.4:23-24 Ap '58.

(Vibrators)

(Vibrators)

NOVOZHILOV, A.I., kand. tekhn. nauk

[Differential equation of the rotation of a body in planeparallel motion; textbook on theoretical mechanics] Differentsial noe uravnenie vrashcheniia tela pri ploskoparallel nom dvizhenii; uchebnoe posobie po teoreticheskoi mekhanike. Magnitogorsk, 1962. 12 p. (MIRA 17:3)

1. Magnitogorsk, Gorno-metallurgicheskiy institut.

L 10673-65 EWT(m)/EWP(e)/EWP(b) Pq-4 AFWL/RAEM(c)/AS(mp)-2/ASD(m)-5/RAEM(1)/SSD/FSD(gs)/ESD(t)/IJP(c) JD/WH

ACCESSION NR: AP4044278

8/0192/64/005/004/0630/0631

AUTHOR: Novozhilov, A. I.; Samoylovich, M. I.; Tsinober, L. I.

TITLE: Short lived paramagnetic centers in germanium doped quarts

SOURCE: Zhurnal strukturnoy khimii, v. 5, no. 4, 1964, 630-631

TOPIC TAGS: quartz, electron paramagnetic resonance, germanium, unpaired electron, x ray irradiation, f ray irradiation, germanium doped quartz

ABSTRACT: Germanium doped quartz, irradiated with x-rays or Y-rays produces an EPR spectrum due to unpaired electrons. In addition to EPR spectrum, resulting from the alkali centers a spectrum is observed which consists of six lines. Their intensity decreases with time. When the temperature of the specimen is lowered to 77 K it is possible to resolve the fine structure consisting of 10 lines, which may be explained by the interaction of electron from an alkali center with the nucleus of Ge<sup>73</sup> isotope. Since splitting is much less than for free germanium atom in the ground state one might conclude that the captured

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L 10673-65

### **ACCESSION NR: AP4044278**

electron spends the majority of time near the oxygen atoms which surround germanium ion. In this work a study was made of the amplitude of the EPR signal as a function of irradiation dose. In the course of 24 hours the amplitude decreases by a factor of 2. The EPR spectrum completely disappears due to heating of the sample to 250C and also after u.v. irradiation. It was found that the rate of disintegration of alkali centers is significantly dependent on the temperature. If at room temperature complete decomposition of centers required several days, at 523 K it requires only several minutes and at 77 K decomposition does not take place at all. Two probable mechanisms are proposed for decomposition of these centers: (1) recombination of electrons with holes which are produced during irradiation of crystals; (2) transition of unstable centers into stable alkali centers due to diffusion of the compensating alkali metal ions. In addition to the above two spectra another spectrum is detected in germanium doped quarts at 77 K. It has no hyperfine structure with g<sub>11</sub> = 1.993±0.002 and g<sub>1</sub> = 1.996±0.002. Orig. art. has: 3 figures.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel skiy institut sinteza mineral

Card 2/3

"APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001237610003-8

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ACCESSION NR: AP40442	78				
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Raw Materials).					
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PI-4 IJF(c) WW/00/AT/WH

ACCESSION NR: AP5016915

UR/0192/65/006/003/0161/0163

AUTHOR: Samoylovich, M. I.; Novozhilov, A. I.

TITLE: Electron spin resonance in irradiated topaz 15

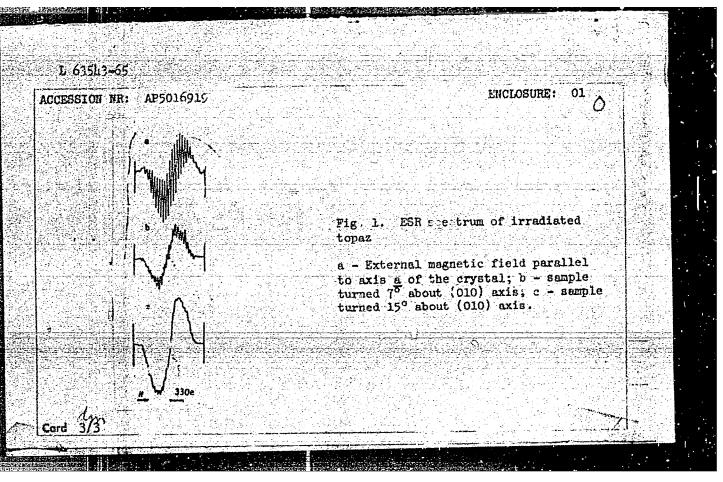
SOURCE: Zhurnal strukturnoy khimii, v. 6, no. 3, 1965, 461 - 466

TOPIC TAGS: topaz, spectrum spin resonance, ESR spectrum, irradiated topaz

ABSTRACT: Several varieties of Volyn topaz were studied by means of electron spin resonance spectra at 77 and 295K. In all natural blown topaz varieties as well as in samples irradiated with K or gamma rays, the ESR spectrum shows a broad isotropic line with g = 1.982 + 0.002 and half-width at 75° with a complex hyperfine structure HFS. The intensity and number of the HFS lines depend on the orientation of the crystal in the magnetic field (see Fig I of the Enclosure). This spectrum is related to the brown color produced by irradiating colorless samples. The spectral line intensity and optical density at first increase symbatically in proportion to the dose, then become saturated at total doses of about 10° reentgen. Both the color and ESR spectrum disappear when the samples are

Card 1/3

ACCESSION NR: AP5016919			4	
heated to 300C. The observed of the mechanism of W. C. Hold conclusion, the authors expres	ton and H. Blum (Phys. Rev. ss their appreciation to A	., 125, 89, 1962). . A. Shaposhnikov, S	"In . V.	
Grum-Grzhimaylo, and Yu. V. Po has: 2 figures and 2 formula:		opaz-samples." -011g	[08]	
ASSOCIATION: Vsesoyuznyy nauc syr'ya, Aleksandrov (All-Union Inorganic Raw Materials)				
SUMMITTED: 080ct64	ENCL: 01	SUB CODE: 1	C NP	1
10 REF SJV: 001	OTHER: 008	ATD INESS:	4049	
7	(1916년) 1912년 대통령 1914년 1월 1일 1일 1일 1일 1일 1일 1일 1924년 1912년 1일			



AUTHOR: Samoylovich, M. I.; Novozkilov, A. I.; Dernov-Pegarev, V. F.; Potkin, L. I.  ORG: All-Union Scientific Research Institute of Synthesis of Mineral Raw Materials, Aleksandrov (Vsesoyuznyy nauchno-issledovatel skiy institut sinteza mineral nogo syr'ya)  TITLE: Electron spin resonance of Mn2+ in molybdates of scheelite structure  SOURCE: Zhurnal strukturnoy khimii, v. 7, no. 1, 1966, 109-110  TOPIC TAGS: manganese, EPR spectrum, molybdate, calcium compound, cadmium compound  ABSTRACT: The ESR spectrum of Mn2+ was studied in single crystals of artificial Ca%OQ, and Ca%OQ, (both of scheelite structure) at 9.4 % at from temporature. Some measurements were made at the temporature of liquid nitrogen. The spin-Hamiltonian constants describing the ESR spectra of Mn2+ in these compounds are tabulated, and compared with those for scheelite. It is noted that the replacement of the anionic groups has practically no effect on the g factor; however, the latter does change slightly when the cations are replaced, the anion being the same. The spin-Hamiltoni- an constant describing the effect of the intracrystalline field of cubic symmetry changes with the anionic groups, but remains virtually unchanged when the cations are replaced. Constant b2, which describes the effect of the intracrystalline field of tetragonal symmetry (the axis of symmetry coincides with the z axis), changes markedly  Card 1/2  UDC: 538.113	ì		-
AUTHOR: Samoylovich, M. I.; Novozhilov, A. I.; Dernov-Pegarev, V. F.; Potkin, I. I.  ORG: All-Union Scientific Research Institute of Synthesis of Mineral Raw Materials, Aleksandrov (Vsesoyuznyy nauchno-issledovatel skiy institut sinteza mineral nogo syr'ya)  TITLE: Electron spin resonance of Mn <sup>2+</sup> in molybdates of scheelite structure  SOURCE: Zhurnal strukturnoy khimil, v. 7, no. 1, 1966, 109-110  TOPIC TAGS: manganese, EPR spectrum, molybdate, calcium compound, cadmium compound ABSTRACT: The ESR spectrum of Mn <sup>2+</sup> was studied in single crystals of artificial CamoQ, and CdmoQ, (both of scheelite structure) at 9.4 m at from temperature. Some measurements were made at the temperature of liquid nitrogen. The spin-Hamiltonian constants describing the ESR spectra of Mn <sup>2+</sup> in these compounds are tabulated, and compared with those for scheelite. It is noted that the replacement of the anionic groups has practically no effect on the g factor; however, the latter does change slightly when the cations are replaced, the anion being the same. The spin-Hamiltoni- an constant describing the effect of the intracrystalline field of cubic symmetry changes with the anionic groups, but remains virtually unchanged when the cations are replaced. Constant bo, which describes the effect of the intracrystalline field of tetragonal symmetry (the axis of symmetry coincides with the z axis), changes markedly	-	L 42887-66 ENT(m)/T/EWP(t)/ETI LIP(c) ID/IG  ACC NR: AP6020384 (A) SOURCE CODE: UR/0192/66/007/001/0109/0110	
All-Union Scientific Research Institute of Synthesis of Mineral Raw Materials, Aleksandrov (Vsesoyuznyy nauchno-issledovatel'skiy institut sinteza mineral'nogo syr'ya)  TITLE: Electron spin resonance of Mn <sup>2+</sup> in molybdatos of scheelite structure  SOURCE: Zhurnal strukturnoy khimii, v. 7, no. 1, 1966, 109-110  TOPIC TAGS: manganese, EPR spectrum, molybdate, calcium compound, cadmium compound ABSTRACT: The ESR spectrum of Mn <sup>2+</sup> was studied in single crystals of artificial Calou, and Calou, (both of scheelite structure) at 9.4 % at from temperature. Some measurements were made at the temperature of liquid nitrogen. The spin-Hamiltonian constants describing the ESR spectra of Mn <sup>2+</sup> in those compounds are tabulated, and compared with those for scheelite. It is noted that the replacement of the anionic groups has practically no effect on the g factor; however, the latter does change slightly when the cations are replaced, the anion being the same. The spin-Hamiltonian constant describing the effect of the intracrystalline field of cubic symmetry changes with the anionic groups, but remains virtually unchanged when the cations are replaced. Constant bo which describes the effect of the intracrystalline field of tetragonal symmetry (the axis of symmetry coincides with the z axis), changes markedly UDC: 538.113		AUTHOR: Samoylovich, M. I.; Novozkilov, A. I.; Dornov-Pegarev, V. F.; Potkin, L. I.	
SOURCE: Zhurnal strukturnoy khimii, v. 7, no. 1, 1966, 109-110  TOPIC TAGS: manganese, EPR spectrum, molybdate, calcium compound, cadmium compound  ABSTRACT: The ESR spectrum of Mn <sup>2+</sup> was studied in single crystals of artificial  Calcot and Cdlot (both of scheelite structure) at 9.4 to at foom temperature. Some measurements were made at the temperature of liquid nitrogen. The spin-Hamiltonian constants describing the ESR spectra of Mn <sup>2+</sup> in these compounds are tabulated, and compared with those for scheelite. It is noted that the replacement of the anionic compared with those for scheelite. It is noted that the replacement of the anionic groups has practically no effect on the g factor; however, the latter does change groups has practically no effect of the anion being the same. The spin-Hamiltonislightly when the cations are replaced, the anion being the same. The spin-Hamiltonian constant describing the effect of the intracrystalline field of cubic symmetry changes with the anionic groups, but remains virtually unchanged when the cations are replaced. Constant b2, which describes the effect of the intracrystalline field of tetragonal symmetry (the axis of symmetry coincides with the z axis), changes markedly UDC: 538.113		ORG: All-Union Scientific Research Institute of Synthesis of Mineral Raw Materials,	
TOPIC TAGS: manganese, EPR spectrum, molybdate, calcium compound, cadmium compound  ABSTRACT: The ESR spectrum of Mn <sup>2+</sup> was studied in single crystals of artificial  Calcot and Cdlot, (both of scheelite structure) at 9.4 lo at from temperature. Some  calcot and constants were made at the temperature of liquid nitrogen. The spin-Hamiltonian  measurements were made at the temperature of liquid nitrogen. The spin-Hamiltonian  constants describing the ESR spectra of Mn <sup>2+</sup> in these compounds are tabulated, and  compared with those for scheelite. It is noted that the replacement of the anionic  groups has practically no effect on the g factor; however, the latter does change  groups has practically no effect on the g factor; however, the latter does change  slightly when the cations are replaced, the anion being the same. The spin-Hamiltoni-  an constant describing the effect of the intracrystalline field of cubic symmetry  changes with the anionic groups, but remains virtually unchanged when the cations are  replaced. Constant b <sub>2</sub> , which describes the effect of the intracrystalline field of  tetragonal symmetry (the axis of symmetry coincides with the z axis), changes markedly		TITLE: Electron spin resonance of Mn2+ in molybdates of scheelite structure	
ABSTRACT: The ESR spectrum of Mn <sup>2+</sup> was studied in single crystals of artificial Calou, and Cdlou, (both of scheelite structure) at 9.4 Mn at from temperature. Some measurements were made at the temperature of liquid nitrogen. The spin-Hamiltonian constants describing the ESR spectra of Mn <sup>2+</sup> in these compounds are tabulated, and compared with those for scheelite. It is noted that the replacement of the anionic groups has practically no effect on the g factor; however, the latter does change slightly when the cations are replaced, the anion being the same. The spin-Hamiltoni-slightly when the cations are replaced, the intracrystalline field of cubic symmetry changes with the anionic groups, but remains virtually unchanged when the cations are replaced. Constant b <sub>2</sub> , which describes the effect of the intracrystalline field of tetragonal symmetry (the axis of symmetry coincides with the z axis), changes markedly		SOURCE: Zhurnal strukturnoy khimii, v. 7, no. 1, 1966, 109-110	
tetragonal symmetry (the axis of symmetry conforms upon 538.113		ABSTRACT: The ESR spectrum of Mn <sup>2+</sup> was studied in single crystals of artificial Callo O4 and Callo O4 (both of scheelite structure) at 9.4 Mo at from temperature. Some measurements were made at the temperature of liquid nitrogen. The spin-Hamiltonian constants describing the ESR spectra of Mn <sup>2+</sup> in these compounds are tabulated, and compared with those for scheelite. It is noted that the replacement of the anionic groups has practically no effect on the g factor; however, the latter does change slightly when the cations are replaced, the anion being the same. The spin-Hamiltonian constant describing the effect of the intracrystalline field of cubic symmetry changes with the anionic groups, but remains virtually unchanged when the cations are	
Card 1/2 UDG: 530.115		tetragonal symmetry (the axis of symmetry conforms	
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with any replacements. For all the crystals, by $\approx 10b_4^0$ , i. e., the surrounding $m^{2+}$ are other than cubic. The ESR spectrum of $m^{2+}$ in CdMoQ, shows lines due to forbidden transitions $\Delta m = \pm 1$ . Authors take this opportunity to thank L. Isinober for his attention to this work. Orig. art. has: 1 table and 1 formula	I.
SUB CODE: 20,07/SUBM DATE: 24Apr65/ OTH REF: 003	
ı	
Card 2/2 bdk)	

ACC NR: AP7000787

(N)

SOURCE CODE: UR/0089/66/021/ .../0360/0363

AUTHOR: Lytkin, V. B.; Troyanov, M. F.; Novozhilov, A. I.

ORG: none

TITLE: Use of the method of calculated losses to choose the characteristics of a

fast reactor

SOURCE: Atomnaya energiya, v. 21, no. 5, 1966, 360-363

TOPIC TAGS: fast reactor, nuclear reactor characteristic, nuclear reactor operation,

plutonium

ABSTRACT: The authors describe a method of using the theoretical fuel-consumption formulas for the analysis of the economics of the fuel cycle of a fast plutonium reactor. The basic formula employed takes care of the initial fuel cost, the efficiency factor, and the fuel component of the cost of electricity generated during the stationary conditions (with credit for fuel reprocessing and for plutonium). The proposed formula also takes into account the prolonged stay of the fissioning material. in the full cycle of the active zone and the screens. Results of calculations are presented, in which the dependence of the fuel component of the calculated expenditures on the heat load and on the "compacting" of the active zone of the reactor of 1000 Mw rated power, is made evident. The formula takes into account the change in the cost of the fuel during the time it stayes in the reactor and time delays in the return of the fuel for reprocessing, and also in the exctraction of the plutonium

Card 1/2

UDC: 338.4: 621.039.516.4

ACC NR: AP6000787

from the screens. Plots are presented of the dependence of the amount of plutonium in the fuel cycle of the active zone and the annual consumption in the fuel elements on the energy load of the active zone, of the dependence of the fuel components of the calculated expenditures and the doubling time on the heat rate, and of the dependence of the fuel component on the ratio of the diameter of the active zone to its height. Examples are presented to show that the method yields a good estimate of the relative roles of the initial investiment in the fuel cycle and the running expenses of the fuel cycle, and consequently makes it possible to choose more correctly the optimal characteristics of the reactor. The authors thank A. I. Leypunskiy and V. V. Orlov for interest in the work and useful discussions, and G. S. Filatov for help with the calculations. Orig. art. has: 3 figures and 3 formulas.

SUB CODE: 18/ SUBM DATE: OlApr66/ ORIG REF: 007

Card 2/2

S/191/60/000/009/008/010 B013/B055

AUTHORS:

Ratner, S. B., Frenkel', M. D., Novozhilov, A. V.

TITLE:

Mechanical Testing of Plastics. 5. Testing of Heat

Resistance

PERIODICAL: Plasticheskiye massy, 1960, No. 9, pp. 69 - 76

TEXT: This publication deals with heat resistance tests of plastics based on the widespread thermomechanical testing methods, i.e., the examination of changes in mechanical properties produced by temperature changes (Figs.1 - 7, Tables 1 - 4). The upper limit of heat resistance of vitrified plastics is the temperature range at which rapid softening occurs. For these plastics the softening point corresponds to the vitrification point  $T_{vitr}$ . With crystalline polymers, the limit of heat resistance is not the  $T_{vitr}$  but practically coincides with the melting point (Ref.1). It is generally known (Ref.2) that the  $T_{vitr}$  is no matter constant since it varies with test conditions. The softening process is strongly affected by the load (Refs.15-17). In the case of some thermo-

Mechanical Testing of Plastics. 5. Testing of S/191/60/000/009/008/010 Heat Resistance B013/B055

plasts, softening was observed to be a linear function of the load

(Refs.15,17). Various thermosetting materials exhibited the same dependence (Figs.2 and 3). It was shown that the softening point drops with increasing load according to  $T = T_0$  - bP, where  $T_0$  = softening point without load, and b = change in heat resistance per unit load. Since  $T_0$  is a characteristic load-independent vitrification point of the material, it must correspond to the vitrification point determined by any method unaffected by other factors, e.g., dilatometrically. This is the case both

unaffected by other factors, e.g., dilatometrically. This is the case both with thermosetting plastics (Fig.4) and thermoplasts. These data show that the dilatometric method may be recommended for testing heat resistance. It must, however, be noted that its lower sensitivity renders it less effective than the method of thermomechanical curves. The most complete characterization of the heat resistance requires determination of  $\mathbb{T}_0$  and b.

For this, tests at 2-3 different loads, at the minimum, are necessary. Industrial methods generally apply only one and the same load (P=const) for testing different types of materials. This results in more or less fortuitous test results which are high for hard materials and low for soft materials. In rapid quality control it is advisable to test heat resistance

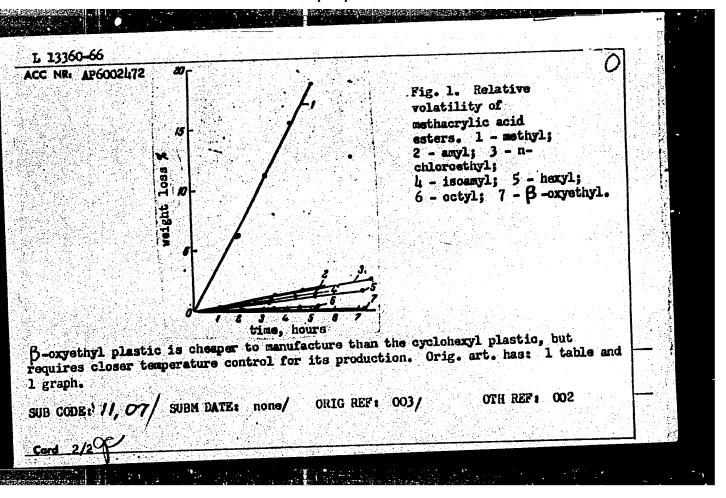
Card 2/3

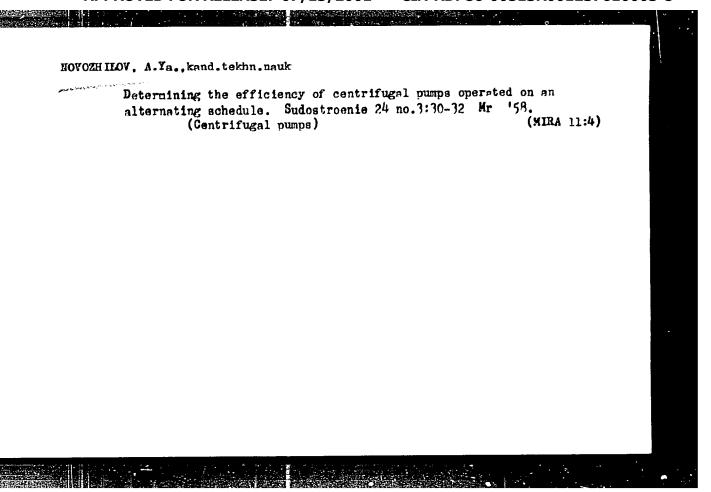
Mechanical Testing of Plastics. 5. Testing of S/191/60/000/009/008/010 Heat Resistance B013/B055

at a load proportional to the initial hardness of the material, i.e., at equal initial deformation ( $\varepsilon_{0}=\mathrm{const}$ ) (Fig.5, Table 2). Widely differing indices are obtained by heat resistance tests under different preset conditions (P = const or  $\varepsilon_{0}=\mathrm{const}$ ) (Figs.6 and 7, Tables 3 and 4). Apart from regulations concerning the general characteristic, the temperature of heat resistance, specifications should also include regulations concerning the heat resistance coefficients of durability and other indices, in accordance with the application of the material or the working conditions the product is to be subjected to. A. P. Aleksandrov is mentioned. There are 7 figures, 4 tables, and 29 references: 23 Soviet, 3 German, 2 US, and 1 Czechoslovakian.

Card 3/3

ACC NR: AP6002472 SOURCE CODE: UR/OL9)/66/000/001/0000/001	
ACC NR. AP6002472 SOURCE CODE: UR/0191/66/000/001/0009/0010  AUTHORS: Novoshilov, A. V.; Rotenberg, I. P.; Vakhtin, V. G.	
ORG: none	
TITLE: Production of polyvinylchloride foam-plastic from a nonvolatile monomer	
SOURCE: Plasticheskiye massy, no. 1, 1966, 9-10	
TOPIC TAGS: polymer, polyvinyl chloride, resin, foam plastic, vinyl plastic, plastic industry/ PKnVl plastic	
ABSTRACT: To prevent the loss of methylmethacrylate and ammonium carbonate, incurred in the conventional manufacture of polyvinyl foam-plastics, the volatility of a number of methacrylic acid esters was determined with the view of selecting less volatile substances. The volatilities (at LCC) of methyl, amyl, n-chloroethyl, isoamyl, hexyl, octyl, and S-oxyethyl were determined, and the experimental results are presented graphically (see Fig. 1). Two foam-plastics were produced, using S-oxyethyl and cyclohexyl methacrylic esters respectively. The ratio of ingredients and the experimental conditions were similar to those used in the industrial manufacture of foam-plastic PKhy-1. Specific volume, strength limit, specific impact viscosity, linear shrinkage, water absorption, and alkali and chloride ion content of the two plastics were determined, and the results tabulated. It is concluded that the	
Cord 1/2 UDC: 678.743.22-496	





SOV/137-59-2-2710

Translation from: Referativnyy zhurnal. Metallurgiya, 1959, Nr 2, p 65 (USSR)

AUTHOR: Novozhilov, B.

TITLE: Some Problems of the Economy of Non-ferrous Metallurgy (Neko-

toryye problemy ekonomiki tsvetnoy metallurgii)

PERIODICAL: Narodnoye kh-vo Kazakhstana, 1958, Nr 4, pp 38-43

ABSTRACT: The author examines the methods of appraisal of nonferrous-metals

deposits. There are two types of appraisals: The preliminary and the comparative. To make a preliminary appraisal it is necessary to know the requirements of the country's national economy for the minerals contained in the ores of the given deposit. For a comparative appraisal it is necessary to know numerous factors determining the technical and economic i character of the contemplated enterprise.

T. K.

Card 1/1

SOV-127-58-8-3/27

AUTHORS:

Pervushin, S.A., Doctor of Economic Sciences, Professor; Novozhilov, B.F.; Zubrilov, L.Ye., Candidate of Technical Scien-

ces

TITLE:

Bases for the Appraisal of Mineral Deposits and Mines (Osnovy otsenki mestorozhdeniy poleznykh iskopayemykh i rudnikov)

PERIODICAL:

Gornyy zhurnal, 1958, Nr 8, pp 18-27 (USSR)

ABSTRACT:

Professor K.L. Pozharitskiy published an article under the abovementioned title, for discussion in Nr 9 (1957) of this periodical. This article is an answer by 3 different authors. Professor S.A. Pervushin says that Pozharitskiy has a wrong conception of the economic appraisal of mineral deposits and mines, because he does not sharply separate the socialist economy from the capitalist economy, considering the problem of appraisel without reference to the basic rules of development of the socialist economy, in particular the Soviet mineral and raw material economy. Pozbaritskiy ignored the experience of Soviet projects institutions as G\_prokhim (A.M. Levin, A.D. Shapiro and others) Giprotsvetmet (Ya.S. Gol'din, Yu.B. Malevskiy, D.D. Ognev) Giproalyuminiy (G.A.Mikhel'son, M.F. Purits), Gipronikel' (L.S. Kul'nitskiy, D.M. Rura), Gipromez (V.O.

Card 1/2

Bases for the Appraisal of Mineral Deposits and Mines SOV-127-58-8-3/27

> Chernyavskiy) and others. The second author, B.F. Novozhilov, bases his criticism on the same lines as the first author. The third author, L.Ye. Zubrilov, finds all propositions made by Pozharnitskiy to be highly controversial. There are 13 references, 11 of which are Soviet and 2 American.

ASSOCIATION: Moskovskiy institut tsvetnykh metallov i zolota (The Moscow Institute of non-ferrous metals and gold). Gosplan Kaz.SSR (The Kazakh SSR Gosplan). Gorno-geologicheskiy institut UFAN (The UFAN Mining-Geological Institute).

1. Minerals--Economic aspects 2. Mines--Economic aspects

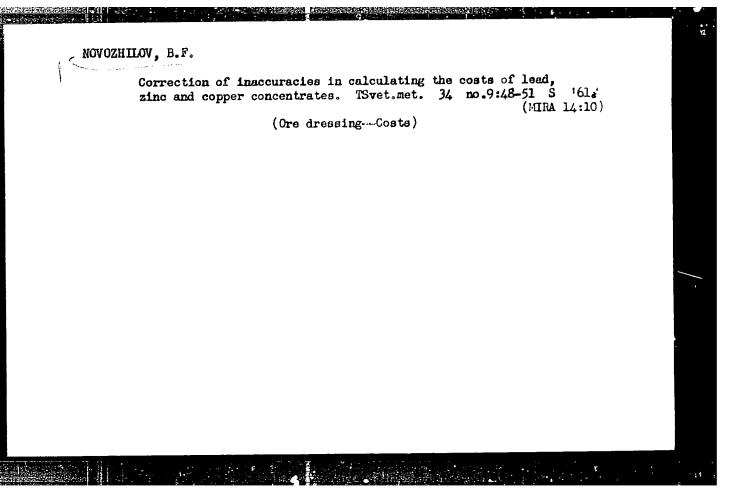
Card 2/2

NOVOZHILOV, B.F.; PARAMONOV, I.V.; ROZHKOV, N.G., red.; KUZEMBAYEVA, A.I.,

[Nonferrous metallurgy in Kazakhstan] TSvetnaia metallurgiia Kazakhstana. Alma-Ata, Kazakhskoe gos. izd-vo, 1960. 34 p.

(MIRA 14:7)

(Kazakhstan—Nonferrous metals—Metallurgy)

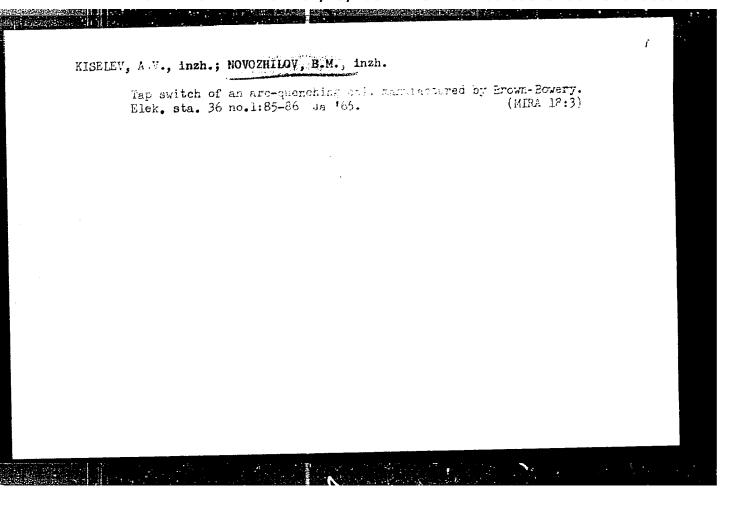


NOVOZHILOV, B.F.

Quality of nonferrous metal ores and the profitableness of production. (MIHA 16:4)

l. Nachal nik planovo-ekonomicheskogo upravleniya Kazakhskogo soveta narodnogo khozyaystva.

(Mining engineering---Costs) (Ores)



Nove: Haler, 18. V.

AUTHOR:

Novozhilov, B.V.

56-5-35/46

TITLE:

Aging Equation for / Quanta (Vozrastnoye uravneniye dlya

/ -kvantov)

PERIODICAL:

Zhurnel Eksperim. i Teoret.Fiziki, 1957, Vol. 33, Nr 5,

pp 1287-1289 (USSR)

ABSTRACT:

Theoretically an equation is derived for the distribution function of multiply scattered f-quanta of low energies (E <<  $m_0c^2$ ), which is analogous to the aging equation for neutrons. It is written down

as follows:  $\partial \varphi(\vec{\tau}, \tau)/\partial \tau = \nabla^{2} \varphi(\vec{\tau}, \tau) - \chi(\tau) \varphi(\vec{\tau}, \tau)$ 

where  $\tau$  denotes the aging of the r-quanta and r  $(\vec{r}, \tau)$  - the moderation density. The spatial distribution and the distribution of the wave length agree satisfactorily with the numerical values of the computed kinetic equation by Goldstein (ref.3). There are 1

figure, 1 table, and 3 references, 1 of which is Slavic.

ASSOCIATION: Institute of Physical Chemistry AN USSS (Institut khimicheskoy fiziki

Akademii nauk SSSR,

SUBMITTED:

June 6, 1957

AVAILABLE:

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NOVOZHILOV, BV

PHASE I BOOK EXPLOITATION

sov/5065

- Leypunskiy, Ovsey Il'ich, Boris Vasil'yevich Novozhilov, and Wevolod Nikolayevich Sakharov
- Rasprostremeniye gamma-kvantov v veshchestve (Propagation of Gamma-Ray Quanta in Matter) Moscow, Fizmatgiz, 1960. 207 p. 6,000 copies printed.
- Ed.: Margulis, U. Ya; Tech. Ed.: Murashova, N. Ya.
- PURPOSE: This book is intended for physicists, engineers, and advanced students concerned with the applications of nuclear physics in industry, as well as with the applications of radio isotopes.
- COVERAGE: The book discusses the theory of Y -quanta propagation and absorption in matter, taking into account the multiple scattering of quanta. It gives the quantitative characteristics (calculated and experimental) of Y-rays for the various representative cases of radiation propagation: propagation in an infinite medium, passage through a layer of finite thickness, reflection from the boundary surface (albedo), etc. Data are given on the attenuation, and the spectral and angular distribution of Y-radiation.

Card 1/4

80099 \$/020/60/131/06/48/071 B011/B005

26.5000

Novozhilov, B. V.

TITLE:

AUTHOR:

Rate of Burning of a 2-Component Mixed Powder Model

PERIODICAL:

Doklady Akademii nauk SSSR, 1960, Vol. 131, No. 6, pp. 1400 - 1403

TEXT: The author describes the process of burning of model powder forming a mechanical mixture of two substances with different properties. Black gunpowder consisting of coal (fuel) and saltpeter (oxidizer) may serve as an example. If the two components strongly differ with respect to their gasification temperatures, and the grains of the more difficultly gasifiable component are sufficiently small, the oxidizer on its gasification will carry the grains along into the gas phase. During their movement, the grains react with the oxidizer which reduces their volume. The concentration of the oxidizer in the gas is also reduced. The heat developed in the reaction is used for heating and accelerating the gas, and for heating and gasifying new portions of solid powder (Refs. 1,2). There is a dependence of the burning rate on pressure and grain size in 2 limiting cases: 1) The burning rate is determined by the diffusion of the oxidizer into the grains (diffusion case); 2) the burning rate is determined by the

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Rate of Burning of a 2-Component Mixed Powder Model \$/020/60/131/06/48/071
B011/B005

kinetics of the chemical reaction (kinetic case). If the burning time of a grain is assumed to be the duration of reaction it appears that in case 1) the burning rate is equal to the square root of pressure, and inversely proportional to the grain size, whereas in case 2) it is inversely proportional to the square root of the grain size. In his paper, the author derives equations (1) - (20) for the absolute burning rate of the powder mentioned in the title. A more rigorous interpretation of physical factors slightly modifies the dependence of the burning rate on pressure and dispersity. Black gunpowder does not fully correspond to the model discussed since the third component - sulfur - is missing. The calculation of the burning rate for black powder and charcoal gives values which are in good agreement with the experimental data. The author thanks 0. I. Leypunskiy for suggesting the subject of the paper, and A. S. Kompaneyets for his advice. There are 5 Soviet references.

ASSOCIATION: Institut khimicheskoy fiziki Akademii nauk SSSR (Institute of

Chemical Physics of the Academy of Sciences, USSR)

PRESENTED: December 29, 1959, by V. N. Kondrat'yev, Academician

SUBMITTED: December 26, 1959

Card 2/2

S/020/61/141/001/018/02<sup>1</sup> B119/B108

11.6200 AUTHOR:

Novozhilov, B. V.

TITLE:

The velocity of the front of an exothermic reaction in a

condensed phase

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 141, no 1, 1961, 151 - 153

TEXT: Proceeding from the theory on the propagation of heat of a flame in gases by Ya. B. Zel'dovich and D. A. Frank-Kamenetskiy (ZhFKh, 12, 100, (1938)) the author computed the velocity of the front of exothermic reactions in a condensed phase as it is the case with polymerization reactions. The velocity u of the reaction front was computed on the reaction that the diffusion coefficient be zero. This was necessary since assumption that the diffusion coefficient coefficient differed very in the example under consideration the diffusion coefficient differed very much from the thermal diffusivity. The relation

 $pp' + \frac{L}{c\kappa} f(T,p) = 0$  was found  $(p(T) = dT/d\kappa, L = heat effect of the reaction, c = specific heat, <math>\kappa$  = thermal diffusivity, T = reaction temperature) In the case of reaction of zeroth order,  $f_0 = Z_0 \exp(-E/RT)$ , the velocity is C and 1/4

The velocity of the front of an...  $\frac{30033}{5/020/61/141,001,018,021}$   $u_0 = \sqrt{\frac{2Z z c \exp(-L/RT_1)RT_1^2}{LE}}$  In chemical reactions of first order,  $f_1 = \frac{Z_1 e^{-ER/T} (1 - \eta)}{LE}$  the velocity of the reaction front is  $u_1 = \sqrt{\frac{Z_1 z c e^{-E/RT_1}RT_1^2}{LE}}$  ( $\eta$  = relative concentration of the reaction product) In the case of an autocatalytic reaction, where  $f = Z_1 e^{-D_1/RT_1} (1 - 1) + Z_2 e^{-E_2/RT} \eta (1 - \eta)$  (9) one finds:

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B119/B108

The velocity of the front of an ...

 $u = \left\{ \frac{Z_{1} \times c \exp\left(-E_{1} / RT_{1}\right) RT_{1}^{2}}{LE \ln\left\{1 + \frac{Z_{1}}{Z_{1}} \exp\left(-\frac{E_{1} - E_{1}}{RT_{1}}\right) \left[\ln\left(1 + \frac{Z_{1}}{Z_{1}} \exp\left(-\frac{E_{1} - E_{1}}{RT_{1}}\right)\right)\right]_{1}^{\frac{E_{1}}{E_{1}} - 1}}\right\}^{1}$ 

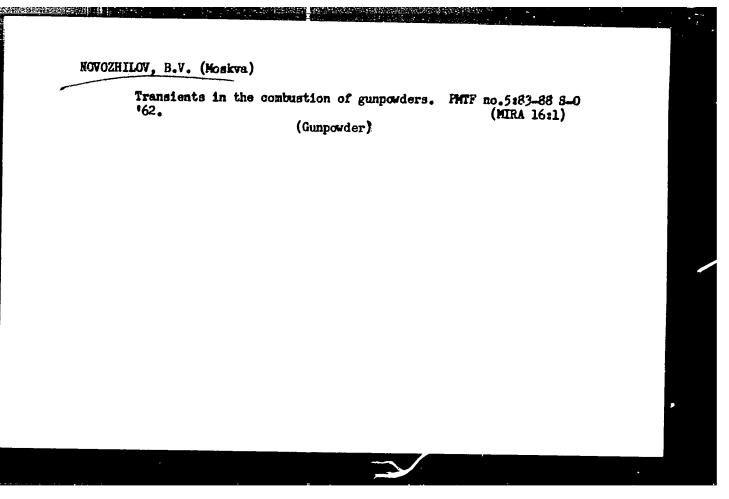
for  $1 + \frac{Z_1}{Z_1} \exp\left(-\frac{E_1 - E_1}{RT_1}\right) \gg e$  и

 $u = \left\{ \frac{Z_{2} \kappa \epsilon e^{-E_{1}/RT_{1}} R T_{1}^{2}}{LE \ln \left[1 + \frac{Z_{2}}{Z_{1}} \exp\left(-\frac{E_{2} - E_{1}}{RT_{1}}\right)\right]} \right\}^{\epsilon},$   $1 + \frac{Z_{2}}{Z_{1}} \exp\left(-\frac{E_{2} - E_{1}}{RT_{1}}\right) \leqslant e.$ 

The author thanks G. B. Manelis for suggestions concerning the work, and A. S. Kompaneyets and E. I. Andriankin for their judgement of the work. There is 1 Soviet reference.

ASSOCIATION: Institut khimicheskoy fiziki Akademii nauk SSSR (Institute of Chemical Physics of the Academy of Sciences USSR)

June 7, 1961, by V. N. Kondrat'yev, Academician PRESENTED: Card 3/4 /2



10016

S/076/62/036/008/007/011 B101/B144

11,7200

AUTHOR:

Novozhilov, B. V.

TITLE:

Theory of burning of a model powder mixture

PERIODICAL: Zhurnal fizicheskoy khimii, v. 36, no. 8, 1962, 1803 - 1806

TEXT: The burning of a powder mixture is studied on the basis of theories advanced by Ya. B. Zel'dovich (Zh. eksperim. i teor. fiz., 12, 498, 1942) and D. A. Frank-Kamenetskiy (Zh. fiz. khimii, 12, 100, 1938) for the case where its components have strongly different thermophysical properties: one component ("fuel") gasifies poorly, the other one ("oxidizer") readily.

Theoretical derivation of the burning rate u gives:  $u \approx 10^{-5}a^{-1}$  p, where a = radius of the fuel particle, p = pressure. The burning rate so found is smaller, however, by several orders of magnitude than that determined experimentally for black powder. This contradicts the assumptions hitherto made (0. I. Leypunskiy, Zh. fiz. khimii, 34, 177, 1960; B. V. Novozhilov, Dokl. AN SSSR, 131, 1400, 1960) that the fuel particles ignite only in the gaseous phase and that inflammation occurs at any gas temperature.

Card 1/2

S/076/62/036/008/007/011 B101/B144

Theory of burning ...

ASSOCIATION: Akademiya nauk SSSR, Institut khimicheskoy fiziki (Academy

of Sciences USSR, Institute of Chemical Physics)

December 27, 1961 SUBMITTED:

Card 2/2

1,2184

5/076/62/036/011/014/021

B101/B180

11.6200

AUTHOR:

Novozhilov, B. V.

TITLE:

The combustion rate of a model powder mixture in the

diffusion range

PERIODICAL:

Zhurnal fizicheskoy khimii, v. 36, no. 11, 1962, 2508-2511

DEXT: The combustion rate, u, is calculated for a powder mixture consisting of two components, A and B, having the same rate of gasification, on the following simplifying assumptions: (1) the powder surface is flat, and lies in the plane x=0; (2) the combustion rate is limited by the diffusion rate, i.e., the rate of the chemical reaction is infinite. It is found that  $u=2D\psi t/\varrho_1 h$ , where D is the diffusion coefficient,  $\varrho$  the gas density,  $\varrho_1$  the density of the solid powder, h half the layer thickness of one component as a measure of the dispersity; t is calculated from the equation

$$\frac{(c/q)(T_1 - T_{ev}) = \tilde{\Phi}(t), \; \tilde{\psi}(t) = (2t/\pi^2 \sum_{n=0}^{\infty} \left\{ 1/(n+1/2)^2 \left[ \sqrt{\pi^2(n+1/2)^2 + t^2} + t \right] \right\}$$

5/076/62/036/011/014/021 B101/B180

The combustion rate of a model...

where c is the specific heat, q the heat effect of the reaction per unit mass of component A, T, the combustion temperature, Tev the evaporation

 $\frac{\pi}{2}(t)$  is given as follows: temperature; n = 0.1,2,... 2.0

3.0 1.0 0.5 2t/π O 0.50 0.45 0.43 0.37

The combustion rate of the powder investigated is a limiting case. In the diffusion range, it is independent of pressure, and is inversely croportional to h the dispersivity. It applied if  $\text{wh}^2/\text{De} \gg 1$ , where w is the rate of the chemical reaction (g/cm²·sec). There are 1 figure and

ASSOCIATION: Akademiya nauk SSSR. Institut knimicheskoy fiziki (Academy of Sciences USSR, Institute of Chemical Physics)

March 18, 1961 SUBMITTED:

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E CHARLET HEREIT B

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\$/020/62/144/006/015/015 B119/B104

11.6366 (also 3619) \_Novozhilov, B. V.

TITLE:

Combustion rate of a mixed powder model

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 144, no. 6, 1962, 1328 - 1330

TEXT: The combustion rate was calculated for a powder made up of two components which react with one another in the gaseous state, emitting heat. For this purpose, both components were assumed to be at the same temperature and to have the same heat of gas formation. The equation

 $n = t \left[ 1 - \left( \frac{\xi}{t_1^2} \right)^3 \frac{t^3}{\alpha_0^2 \varphi_0 (\varphi_0 + 2t)^2 \varrho^{\varphi_0/t}} \right]$ of the reset was obtained. The relation

of the reaction rate to the rate  $\xi = Eo/Rq$  of mixing by diffusion is

characterized by  $n = \frac{n^2 oz}{D} \left(\frac{\tau_1^2}{\xi}\right)^3 \exp(-\xi/\tau_1)$ , and  $n \sim t$  when  $n \ll 1$ . The combustion rate is then proportional to the gas pressure and independent of dispersity, in accordance with Ya. B. Zel'dovich's equation (ZhETF, 12, Card 1/2 . . .

S/020/62/144/006/015/015 B119/B104

Combustion rate of a mixed ...

498 (1942)), but when the values of n are higher, the combustion rate is independent of n and therefore also independent of the gas pressure which is proportional to n; it is inversely proportional to dispersity. There are 2 figures.

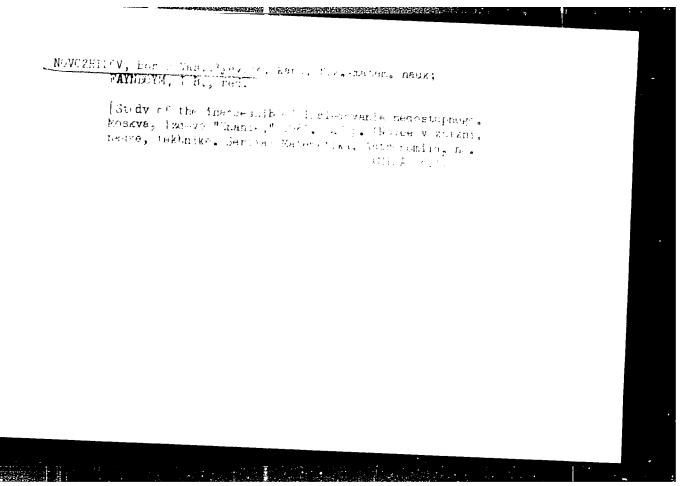
ASSOCIATION: Institut khimicheskoy fiziki Akademii nauk SSSR (Institute

of Chemical Physics of the Academy of Sciences USSR)

PRESENTED: February 6, 1962 by V. N. Kondrat'yev, Academician

SUBMITTED: January 30, 1962

Card 2/2



NOVOZHILOV, Boris Vasil'yevich; SHUSTOVA, I.B., red.; ATROSHCHENKO,
L.Ye., tekhn. red.

[Solid liquids] O tverdykh zhidkostiakh. Moskva, Izd-vo
"Znanie," Moskva, 1963. 47 p. (Narodnyi universitet kul'tury:
"Znanie," Hoskva, 1963. 47 p. (Narodnyi universitet kul'tury:
Estestvennonauchnyi fakul'tet, no.5)

(Solids) (Liquids)

ACCESSION NR: AP4013336

S/0020/64/154/003/0690/0691

AUTHOR: Novozhilov, B. V.

TITLE: Temperature dependence of kinetic characteristics of exothermic reactions in the condensed state

SOURCE: AN SSSR. Doklady\*, v. 154, no. 3, 1964, 690-691

TOPIC TAGS: kinetic characteristics, activation energy, temperature dependence, exothermic reactions, explosives

ABSTRACT: It has been found in the study of slow chemical transformations of explosives that the preexponential factor and the activation energy increase with temperature (K. K. Andreyevand A. F.Belyayev, Theory of Explosives, 1960). This increase had been attributed to the existence of chains—the length of which increases with temperature. It is assumed in the present paper that these chains are initiated by the thermal reaction of the nonreacted molecules located in the vacinity of the reacted molecule. The probability \*\*forsuch initiation is

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 ACCESSION NR: AP4013336

calculated. It is shown that with properly selected x the effective values of the activation energy and of the preexponential factor are of the observed order of magnitude and show the observed temperature dependence. "The author is grateful to A. S. Kompaneyets for & discussion." Orig. art. has: no figures

ASSOCIATION: Institut khimicheskoy fiziki Akademii Nauk SSSR (Institute for Chemical Physics AN SSSR)

SUBMITTED: 12Sep63

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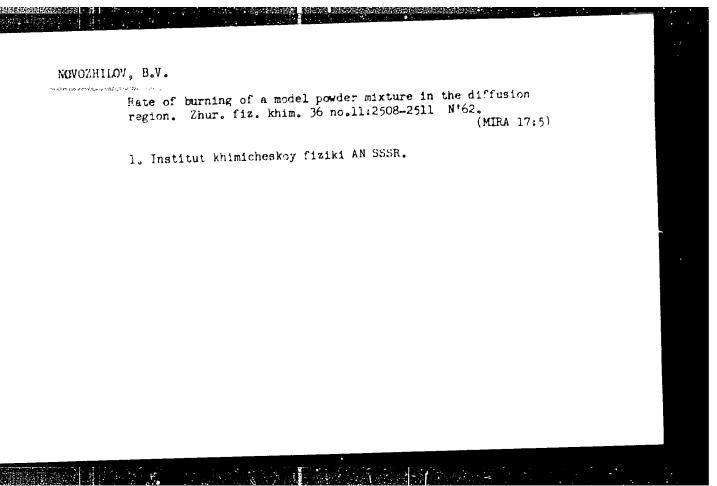
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Card 2/2



 ACCESSION NR: AP4041203

s/0207/64/000/003/0139/0144

AUTHORS: Istratov, A. G. (Moscow); Librovich, V. B. (Moscow); Novozhilov, B. V. (Moscow)

TITLE: Concerning the approximation method in the theory of uneven combustion rate of a powder

SOURCE: Zhurnal prikladnoy mekhaniki i tekhnicheskoy fiziki, no. 3, 1964, 139-144

TOPIC TAGS: combustion rate, combustion stability, computer result, combustion temperature, temperature gradient

ABSTRACT: Analytical expressions for an uneven combustion rate were derived for, a powder model with a combustion rate dependent only on the pressure and surface temperature gradient of the condensation phase. Instantaneous and exponential pressure variations were studied. The steady powder combustion rate was investigated for both the linear and exponential dependence on the initial powder temperature. In steady combustion the rate is determined by the initial temperature  $T_0$  and the pressure p, and a relation exists between  $T_0$  and the temperature gradient at the boundary of the condensation phase p. Knowing this,  $T_0$  was found as a

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ACCESSION NR: APLOL1203

function of p and  $\Phi$ , and the combustion rate was expressed in these parameters. Ya. B. Zel'dovich (O skorosti goreniya porokha pri peremennom davlenii. PMF, 1964, No. 3) showed that this could also be done for uneven burning, but in this case  $\varphi$ must be determined from the solution of the thermal conductivity equation in the solid phase. The problem was worked out with the dimensionless variables; it consisted of finding functions determining the uneven combustion rate and the temperature distribution in a solid phase. This had been previously done by a computer using the approximation method of integral equations. The uneven combustion rate was studied for a linear dependence of the powder combustion rate on the initial temperature. For the purpose of illustrating the derived results, uneven combustion rates with a sharp and an exponential decrease of pressure were examined by the approximation method and compared to computer results with satisfactory agreement. Extinguishing of the powder may take place with a rather rapid decrease in the pressure, and an instantaneous decrease leads to a negative radical which is unsolvable. The final portion of the paper is devoted to the study of the uneven combustion rate with an exponential dependence of the powder combustion rate on the initial temperature. The authors thank 0. I. Leypunskiy and 0. I. Barenblatt for their critique and advice. Orig. art. has: 6 figures and 28 equations.

ASSCCIATION: none

Card 2/3

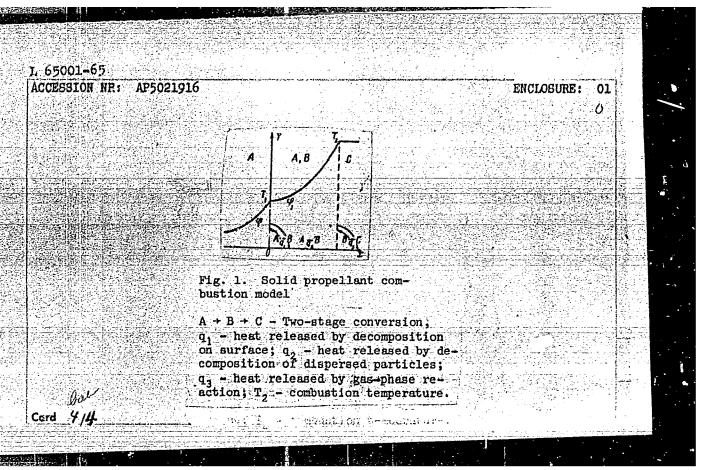
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SOURCE: Zhurnal prikladnoy mekhaniki i tekhnicheskoy fiziki, no. 4, 1965, 157-160  SOURCE: Zhurnal prikladnoy mekhaniki i tekhnicheskoy fiziki, no. 4, 1965, 157-160  1071C TAGS: combustion, solid propellant, combustion stability, combustion instability, stability criterion  ABSTRACT: An analysis of solid propellant combustion was made on the basis of a model which was assumed to have a variable propellant surface temperature and a two-stage combustion process during which the propellant is first decomposed and gasified and then reacts in the gas phase to yield the final products (see Fig. 1 of Enclosure). It is assumed that the reactions in the gas and condensed phases take place without inertia. The heat release due to decomposition of the condensed and dispersed phases and that of the gas-phase reaction are considered. Based on the concept that combustion is either stable or unstable when small perturbations of the burning velocity or surface temperature are either damped or amplified, the following stability criteria were derived. Combustion is always stable when k < 1; when	ACCESSION NR: AP5021916	<b>ਪਜ/0207/65/000/004/0157/0160</b>	
TOPIC TAGS: combustion, solid propellant, combustion stability, combustion instability, stability criterion  ABSTRACT: An analysis of solid propellant combustion was made on the basis of a model which was assumed to have a variable propellant surface temperature and a two-stage combustion process during which the propellant is first decomposed and gasified and then reacts in the gas phase to yield the final products (see Fig. 1 of Enclosure). It is assumed that the reactions in the gas and condensed phases take place without inertia. The heat release due to decomposition of the condensed and dispersed phases and that of the gas-phase reaction are considered. Based on the concept that combustion is either stable or unstable when small perturbations of the burning velocity or surface temperature are either damped or amplified, the following	SOURCE; Zhurnal prikladnoy mekhaniki i tekhnich	· 즐겁게 하고 있었다. 그렇다 나는 그 그리고 하다 다른 개요하다.	
stage combustion process during which the propellant is first decomposed and gasified and then reacts in the gas phase to yield the final products (see Fig. 1 of Enclosure). It is assumed that the reactions in the gas and condensed phases take place without inertia. The heat release due to decomposition of the condensed and dispersed phases and that of the gas-phase reaction are considered. Based on the concept that combustion is either stable or unstable when small perturbations of the burning velocity or surface temperature are either damped or amplified, the following	TOPIC TACS: combustion, solid propellant, combustability, stability criterion  ABSTRACT: An analysis of solid propellant combusts.	کارلیار) Justion was made on the basis of a	
dispersed phases and that of the gas-phase reaction are considered. Based on the concept that combustion is either stable or unstable when small perturbations of the burning velocity or surface temperature are either damped or amplified, the following	model which was assumed to have a variable proper stage combustion process during which the propel fied and then reacts in the gas phase to yield the Enclosure). It is assumed that the reactions in	llant surface temperature and a two- lant is first decomposed and gasi- the final products (see Fig. 1 of the gas and condensed phases take	
	dispersed phases and that of the gas-phase read concept that combustion is either stable or unst huming velocity or surface temperature are eith	able when small perturbations of the ter damped or amplified, the following	

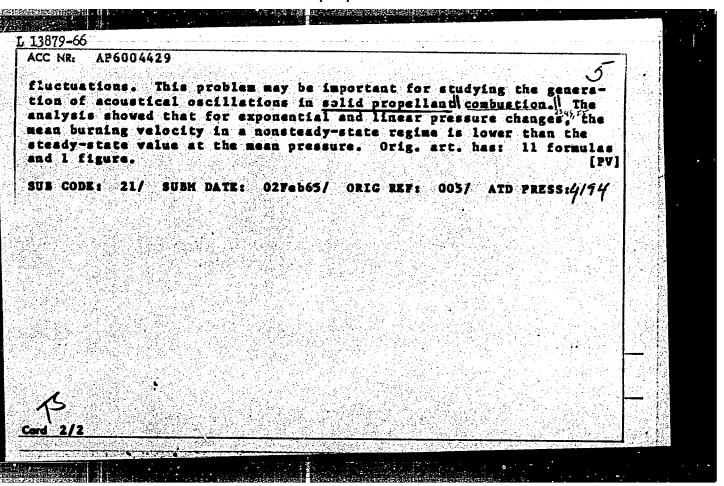
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here,			
	$\left(\begin{array}{c} \dot{K} = (T_1 - T_0) \left(\frac{\partial \ln m}{\partial T_0}\right)_p \end{array}\right)$		
nd			
	$i_{r} = (\partial r_{\perp} / \partial r_{0})_{p} I_{r}$		
malysis only $m(\Psi, p)$ must be known	; \$\temperature profile; Ti, surface temper rature. It is noted that for steady-state sown, while under nonsteady-state conditions art. has: 3 formulas and 2 figures.	ature; tability also [PV]	
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"APPROVED FOR RELEASE: 07/13/2001 CIA-RDP86-00513R001237610003-8

L 65001-65 ACCESSION NR: AP5021916					
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Card 3/1			5.시간 이 1. 전환 7.11 이 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1.		



L 13879-66 EWT(m)/FBA/ETC(m)-6/EWP(f) WW/JWD AP6004429 SOURCE CODE: UR/0414/65/000/003/0041/0044 ACC NR AUTHOR: Novozhilov, B. V. (Hoscow) ORG: none TITLE: The mean burning velocity during hermonic pressure chang SOURCE: Fisika goreniya i veryva, no. 3, 1965, 41-44 TOPIC TAGS: combustion instability, solid propellant, propulsion ABSTRACT: During slow pressure variation, the temperature profile of a propellant follows the pressure change and the burning velocity has a quasi steady-state value. When the pressure changes rapidly, the temperature profile lags behind the pressure change and the burning velocity assumes a nonsteady-state value. Novozhilov (PMTF, 1962, 5.) and Istratov, Librovich, and Novoshilov (PMTF, 1964, 3.) investigated the latter regime by analyzing the effect of instantaneous or exponential pressure changes or changes caused by a sudden nozzle constriction in a solid propellant motor. In the present study, a model is considered in which the burning velocity of the propellant is a function of the pressure and the temperature gradients on the solid surface only. Expressions were derived for evaluating the deviation of the pressure from the steady-state value during harmonic pressure 532.46 Card 1/2



## "APPROVED FOR RELEASE: 07/13/2001 C

#### CIA-RDP86-00513R001237610003-8

UR/0207/65/000/006/0141/0144 SOURCE CODE: ACC NR. AP6002372 Novoshilov, B. V. (Moscow) ORG: none TITLE: Powder combustion during harmonic pressure changes Zhurnel prikladnoy mekhaniki i tekhnicheskoy fiziki, no. 6, 1965, 141-144 SOURCE: TOPIC TAGS: combustion instability, solid propellant, propellant, combustion, burning velocity ABSTRACT: A theoretical analysis of solid propellant combustion showed that a solid propellant constitutes an oscillating system with a natural frequency and a damping decrement. The model was based on the assumption that the surface temperature Tg depends on the pressure p and the initial temperature  $T_0$ . The relationship for the steady state burning velocity  $m(T_0,p)$  was then transformed to the relationship for the nonsteady-state burning velocity m(f,p), where f is the temperature gradient at the surface. Relationships in terms of k, r, and m, where **Card** 1/3

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ACC NR: AP6002372

It was shown that a stable regime characterized by k>1 and  $r \ge (k-1)^2/(k+1)$  can be reestablished by relaxation of the nonsteady-state temperature distribution. The time exponent characterizing the approach to the steady-state regime has the form  $\exp[2(u^{\circ})^2 \cdot t/\kappa]$ , where  $u^{\circ} = u^{\circ}/\rho$  (burning velocity), and

$$\Omega = \frac{(k-1)^{p}-r(k+1)}{2^{p}} \pm i \frac{k-1}{2^{p}} \sqrt{(2k-r+2)r-(k-1)^{p}}.$$

 $\chi = - \operatorname{Re} \Omega$  is the decrement which characterizes damping of the oscillations.

characterizes the oscillation frequency. The character of the oscillating regime depends on the ratio  $\lambda/\omega$ . When  $\lambda<\omega$ , the damping is small. As r approaches  $(k-1)^2/(k+1)$ , i.e., at the stability limit,  $\lambda\to 0$ . Thus, it can be seen flatwhen k>1, the propellant is an oscillating system with a given frequency and damping decrement. When k<1, the steady-state regime is approached aperiodically, i.e., without passing through the steady-state value. Expressions for the burning velocity in the presence of forced harmonic oscillations were also derived and conditions for resonance sence of forced harmonic oscillations were also derived and conditions characteriziere analyzed. It was shown that the equations and boundary conditions characterizing resonance are nonlinear, so that all phenomena related to nonlinear oscillations such as ambiguous relationships between the burning velocity and the pressure fluctu-

Card 2/3

	O	
ation frequency, sudden transitions between regimes, and resonance at frequency can be expected. Orig. art. has: not coinciding with the natural frequency can be expected. Orig. art. has: tions.	encies 25 equa- [PV]	
SUB CODE: 21/ SUBM DATE: 23Jun65/ ORIG REF: 006/ ATD PRESS: 4/97		

NOVOZHILOV. Boris Vasil'yevich, kand. fiz.-matem. nauk; FOYNEYV,

I.B., red.

[Monte Carlo method] Metod Monte-Karlo. Moskva. Znanie,
[1966. 46 p. (Novoe v zhizni, nauke, tekhnike IX Seriia:
Fizika, Matematika, Astronomiia, no.3) (MIRA 19:1)

ACC NR. AP7000048

SOURCE CODE: UR/0207/66/000/005/0031/0041

AUTHOR: Novozhilov, B.V. (Moscow)

ORG: none

TITLE: Nonlinear fluctuations in the burning velocity of powder

SOURCE: Thurnal prikladnoy mekhaniki i tekhnicheskoy fiziki, no.5, 1966, 31-41

TOPIC TAGS: burning velocity, combustion, combustion instability, solid propellant

ABSTRACT: A theoretical analysis was made of nonlinear burning velocity fluctuations in a solid propellant. The model assumed that during harmonic pressure changes the surface temperature of the propellant depends on the pressure and the initial temperature. It was shown that due to the nonlinearity of the heat conduction equation and the dependence of the burning velocity and surface temperature on the pressure and initial temperature, the fundamental frequency of the burning velocity fluctuation changes and higher harmonics of the burning velocity and temperature fluctuations appear. Resonance curves were constructed for the case of nonlinear resonance when the frequency of the pressure fluctuations is close to the natural frequency of the

Card 1/2

#### ACC NR: AP7000048

The problem of stability in given sections of the propellant. resonance curves must be solved by special investigations because unlike electrical and mechanical systems in which stability conditions are described by ordinary differential equations, in the case of combustion instability, nonlinear partial differential equations are involved with nonlinear relationships between the burning velocity, the temperature, and the temperature gradient at the surface. It was shown that the allowance for nonlinear effects under resonance conditions leads to a dependence of the frequency and the damping decrement on the amplitude of fluctuations. It was shown that self-oscillating combustion regimes are possible at a constant pressure. The author thanks Ya. B. Zel'dovich, G. I. Barenblatt, A. S. Kompaneyts, and O. I. Leypunskiy for discussions of and advice on the investigated problems. Orig. art. has: 59 [WA-68] formulas and 5 figures.

SUB CODE: 21/ SUBM DATE: 02Jun66/ ORIG REF: 007/ OTH REF: 001

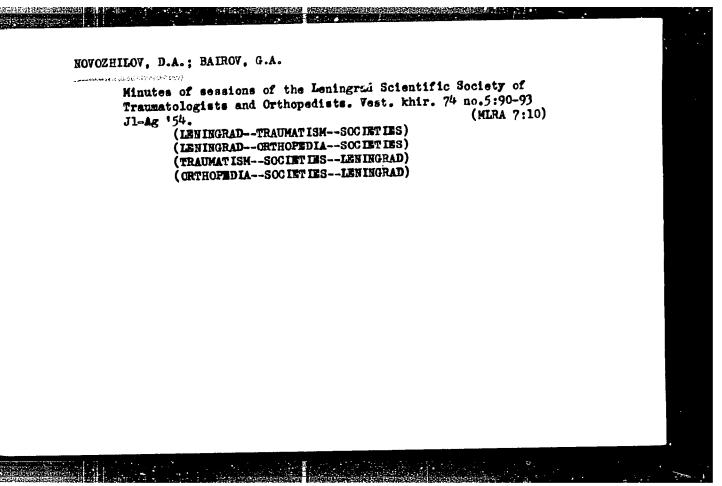
'Card 2/2-

NOVOZHILOV, D.A. prof. (Leningrad, P-183, Naberezhnaya Chernoy rechki, d.24. kv.39)

Pain and its significance in injuries. Ortop. travm. i protes. 24 no.6:3-11 Je \*63 (MIRA 16:12)

Traumatic Myogelosis as a Complication in Hip Fractures"

Trudy Voyanno-Morskoy Med Akad., Vol I, 1948.
S461

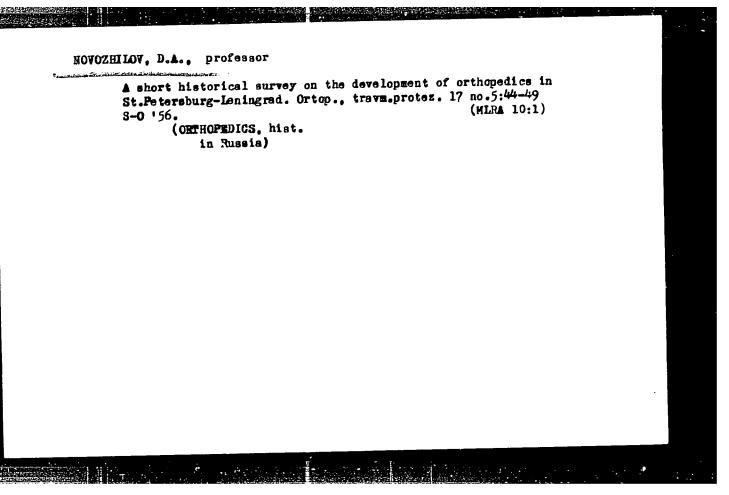


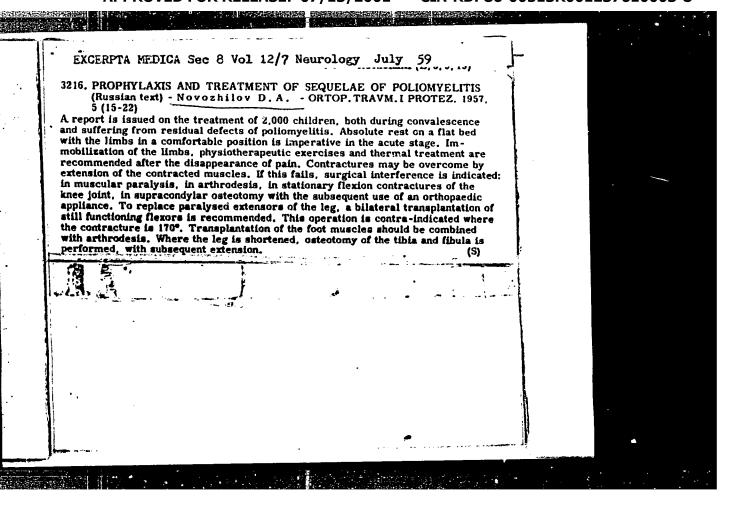
 NOVOZHILOV, D.A., professor

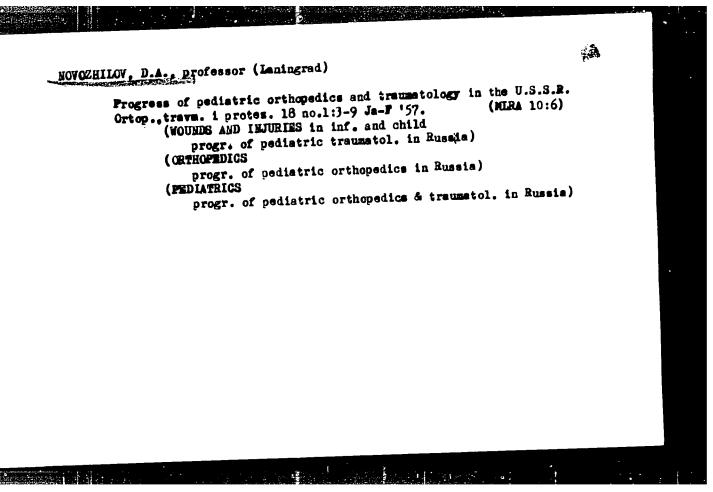
Professor S.A. Hovotel'nov. Ortop., travm. i protez. 17 no.4:70-71
Jl-Ag '56. (MLRA 9:12)

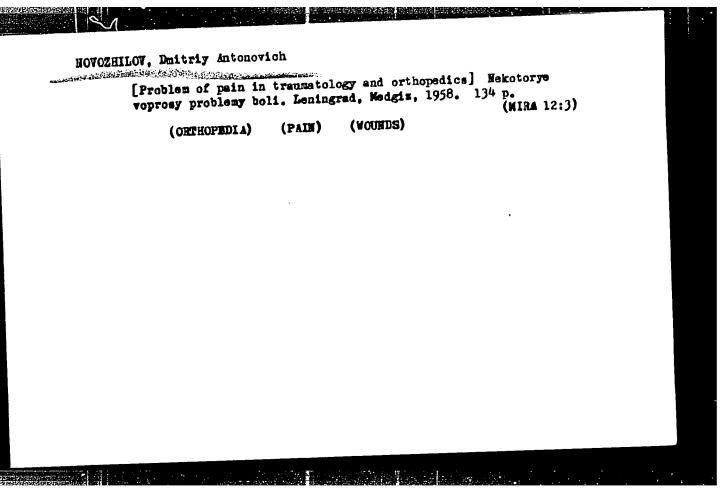
1. Predsedatel' Leningradskogo obshchestva ortopedov i travmatologov.

(HOVOTEL'NOV, SERGEI AHRAMOVICH)









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MIKHAYLOVSKAYA, L.A., kand.biol.neuk, NOVOZHILOV, D.A., mrof. IVANOV, I.I., mrof.

Biochemical studies of the muscle in poliomyelitis and their significance for the clinician. Ortop.travm.i protez. 19 no.3:28-32 My-Je '58 (MIRA 11:7)

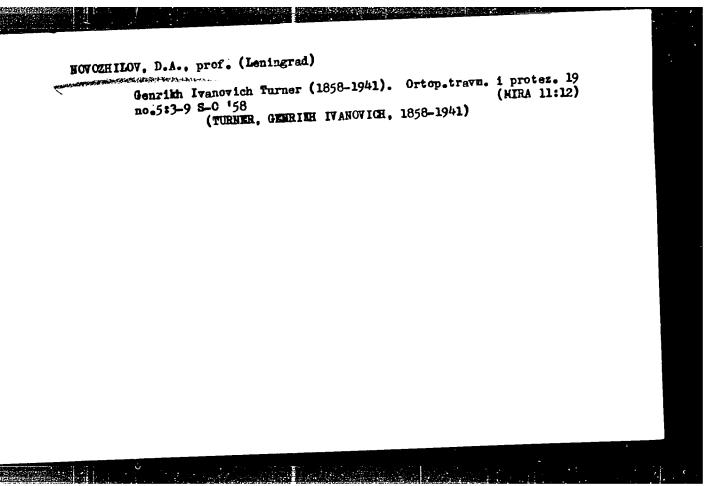
1. Iz nauchno-isaledovrtel'skogo detskogo ortopedicheskogo instituta im. G.I. Turnera i kafedry biokhimii Leningridskogo mediatricheskogo meditainskogo instituta.

(POLIOMYELITIS, nathol.

musc., biochem. changes (Rus))

(MUSCLE, nathol.

in poliomyelitis, biochem. changes (Rus))
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IVANOV. I.I.; YUR'YEV, V.A.; NOVOZHILOV, D.A.; MIKHAYLOVSKAYA, L.A.;

KRYMSKAYA, B.M.

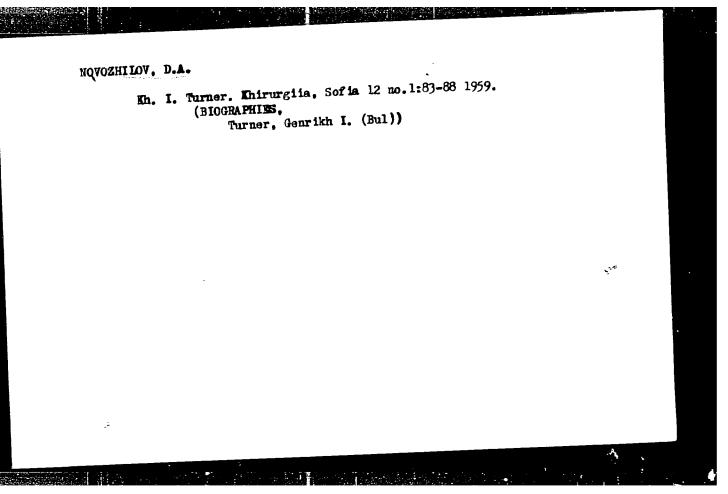
Biochemical determination of the functional condition of muscles in poliomyelitis. Vop.med.khim. 5 no.4:243-250 Jl-Ag \*59.

(MIRA 12:12)

1. Kafedra biokhimii Leningradskogo pediatricheskogo meditainskogo instituta i biokhimicheskaya laboratoriya Nauchno-issledovatel\*skogo detskogo ortopedicheskogo instituta imeni G.I. Turnera.

(POLIOMYELITIS pathol.)

(MUSCLE PROTEINS)



MOVOZHILOV, D.A., prof.

A further development of medical science. Ortop.travm. 1
protez. 2C no.3:3-6 Mr '59. (MIRA 12:6)

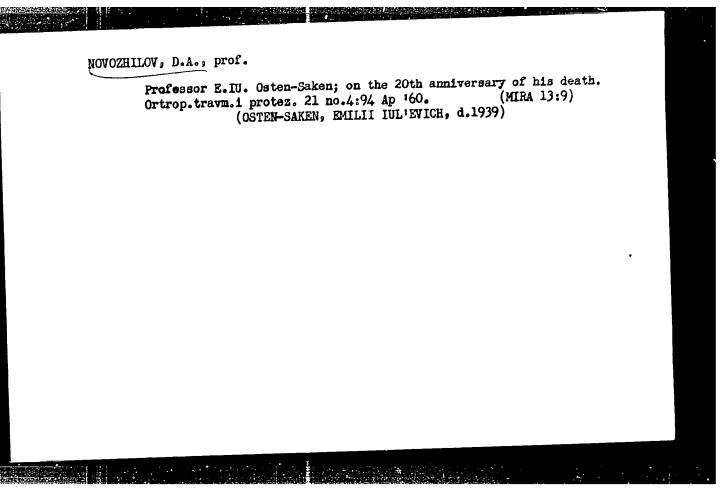
1. Predsedatel' pravleniya Leningradskogo obshchestva
travmatologov i ortopedov.
(MEDICINE
in Russia (Rus))

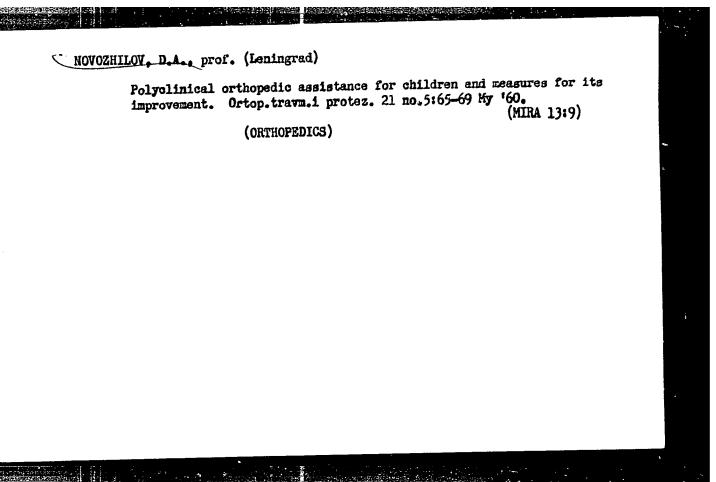
# MOVOZHILOV, D.A., prof.

Basic questions in the organization of treatment of poliomyelitis in children. Ortop.travm.i protez. 20 no.8:3-8 Ag 159. (MIRA 12:11)

1. Iz Nauchno-issledovatel skogo detskogo ortopedicheskogo instituta im. G.I. Turnera (dir. - prof. M.N. Goncharova).

(POLIOMYELITIS, therapy)





#### NOVOZHILOV, D.A., prof.

Cradle of children's orthopedics, the G.I. Turner Institute (on the 70th anniversary of its founding). Ortop.travm.i (MIRA 14:8) protez. no.6:76-84 '61.

1. Iz Detskogo ortopedicheskogo instituta im. G.I. Turnera (dir. - prof. M.N. Goncharova, zam. direktora - prof. D.A. Novozhilov).

(ORTHOPEDICS—HOSPITALS AND INSTITUTIONS)

## NOVOZHILOV, D.A., prof.

Pediatric orthopedic institution. Vop. okh. mat. i det. 6 no.5: (MIRA 14:10)

1. Zamestitel direktora Detskogo ortopedicheskogo instituta imeni G.I:Turnera po nauchnoy rabote (direktor - prof. M.N.Goncharova). (ORTHOPEDIA-HOSPITALS AND INSTITUTIONS)

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NOVOZHILOV, D. A., prof. (Leningrad 187, naberezhnaya fontanki, d.2, kv. 322-a

Pathogenesis of cerebral paralysis in childhood, Ortop., travm. i protez. 22 no.8:3-10 Ag 161. (MIRA 14:12)

1. Iz Gosudarstvennogo nauchno-issledovatel<sup>†</sup>skogo detskogo ortopedicheskogo instituta im. G. I. Turnera (dir. - prof. M. N. Goncharova).

(CEREBRAL PALSIED CHILDREN)

 NEMIRA, V. G., kand. med. nauk; NOVOZHILOV, D. A., prof.

Vitamin requirement of children in the rehabilitation period following poliomyelitis. Ortop., travm. i protez. 22 no.8:34-36 Ag '61.

(MIRA 14:12)

1. Is Gosudarstvennogo nauchno-issledovatel skogo detskogo ortopedicheskogo instituta im. G. I. Turnera (dir. - prof. M. N. Goncharova)

(POLICMYELITIS) (VITAMIN METABOLISM)

NOVOZHILOV, Dmitriy Antonovich, prof.; OGIY, I.A., red.; KHARASH,
G.A., tekhn. red.

[Poliomyelitis] Poliomielit. Leningrad, Medgiz, 1962. 33 p.
(MIRA 15:6)

(POLIOMYELITIS)

NOVOZHILOV, E.A., prof. (Leningrad, P-183, naberezhnaya Chernoy rechki, d.21, kv.39)

Defective posture and the so-called idiopathic scoliosis.
Ortop. travm. 1 protez. 26 no.6174-79 Je '65.

(MIRA 18:8)

NOVOZHILOV, Dmitriy Antonovich, prof.; SELIVANOV, Ye.F., red.

[G.I.Turner's remarkable life] Zamechatel'naia zhizn'
G.I.Turnera. Leningrad, Meditsina, 1965. 142 p.
(MIRA 18:10)

NOVOZHILOV, D.I.

State standard 2386-62 for level amposites. Izm. tekt. no.l:
18-19 Ja '64.

(MIRA 17:11

GUMENSKIY, Boris Mikhaylovich, prof.; NOVOZHILOV, Gennadiy Fedorovich, assistent; KCVRIZHNYKH, L.P., red.; DONSKAYA, G.D., tekhn. red.

[Thixotropy of soil and its calculation in the construction of roads and road bridges] Tiksotropiia gruntov i ee uchet pri stroitel'stve avtomobil'nykh dorog i mostov. Moskva, Nauchnotekhn.izd-vo M-va avtomobil'nogo transp. i shosseinykh dorog RSFSR, 1961. 106 p. (MIRA 15:2)

(Soil mechanics—Research) (Road construction)
(Bridge construction)

GUMENSKIY, B.M. (Leningrad); NOVOZHILOV, G.F. (Leningrad)

Increase in the bearing capacity of piles during the "resting" process. Osn., fund. i mekh. grun. 3 no.4:16-17 '61. (MIRA 14:8)

(Piling (Civil engineering))

SOKOLOV, B. G., inzh.; NOVOZHILOV, G. F., inzh.

Means for preventing the freezing of clay to the surface of metal and wood. Stroi. mat. 8 no.9:37-38 S '62.

(MIRA 15:10)

(Clay)

NOVOZHILOV, G.F., inzh.

Frost heave on bridge approaches and its prevention. Transp. stroi. 12 no.10:50-51 0 '62. (MIRA 15:12) (Roads—Frost damage)

MOVOZHILOV, G.F., assistent

Heaving of the roadbed at the approaches to engineering structures. Shore trud. LIZHT no.203:21-20 '63.

Some regularities in the increase of pile capacity with time. Thid.:63-70 (MIHA 18:8)

SOKOLOV, B.G., inzh.; NOVOZHILOV, G.F., assistent

Laboratory analysis of the action of a special composition proventing the freezing of clays to wood and metal surfaces.

Shor. trud. LIZHT no.203:71-81 \*63. (MIRA 18:8)

NOVOZHILOV, Grigoriy Gerasimovich, slesar'; YARTSEV, N., red.;
KUZNETSOVA, A., tekhn. red.

[We mechanize the work of builders]Mekhaniziruem trud stroitelei. Moskva, Mosk. rabochii, 1962. 30 p. (MIRA 15:9)

1. Moskovskiy zhilishchno-stroitel'nyy trest (for Novozhilov).
(Building-Technological immovations)

NOVOZHILOV, G.I., kand. tekhn. nauk; CHIZHOV, A.T., kand. tekhn. nauk.

New working methods in assembling yards. Transp. stroi. 7 no.11:
27-28 N '57.

(Loading and unloading) (Railroads--Construction)

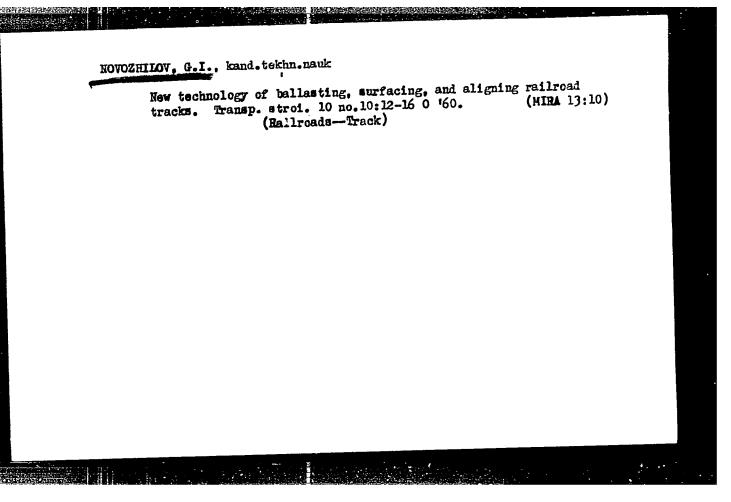
HOVOZHILOV, G.I., kand. tekhn. nauk

Use of reinforced concrete ties abroad. Transp. stroi. 9 no. 5:52-54
Ap '59.

(Railroads-Ties, Concrete)

MOVOZHILOV, G.I., kand.tekhn.nauk

Thirty years of the Turkestan-Siberia Railroad. Transp.stroi.
10 no.5162 by '60. (MEA 13:7)
(Soviet Central Asia--Railroads--Contruction)



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Track skeletonizing since in the concrete bile. Thus, stroid (Minimum) (Mini
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NOWOZHILOV, G.I., kand.tekhn.nauk

New technique for laying the second track with reinforced concrete crossties. Transp. stroi. 12 no.9:13-16 S '62. (MIRA 16:2)

(Railroads—Track)

